

(19)



Europäisches Patentamt

European Patent Office

Office européen des brevets

(11)

EP 0 950 070 B1

(12)

EUROPEAN PATENT SPECIFICATION

(45) Date of publication and mention
of the grant of the patent:
06.02.2002 Bulletin 2002/06

(51) Int Cl.7: **C08F 2/20, C11D 17/00,
A01N 25/10**

(21) Application number: **97950317.4**

(86) International application number:
PCT/GB97/03531

(22) Date of filing: **23.12.1997**

(87) International publication number:
WO 98/28339 (02.07.1998 Gazette 1998/26)

(54) PARTICLES HAVING SURFACE PROPERTIES AND METHODS OF MAKING THEM

TEILCHEN MIT OBERFLÄCHENEIGENSCHAFTEN UND VERFAHREN ZU IHRER HERSTELLUNG
PARTICULES PRESENTANT DES PROPRIETES DE SURFACE ET PROCEDES DE PRODUCTION
DE CES PARTICULES

(84) Designated Contracting States:
**AT BE CH DE DK ES FI FR GB GR IE IT LI LU NL
PT SE**

(30) Priority: **23.12.1996 EP 96309466**

(43) Date of publication of application:
20.10.1999 Bulletin 1999/42

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(56) References cited:
EP-A- 0 441 512 EP-A- 0 604 109
WO-A-92/18222 WO-A-93/22417

- **DATABASE WPI Section Ch, Week 8612 Derwent
Publications Ltd., London, GB; Class A14, AN
86-079085 XP002057513 & JP 61 028 441 A (LION
CORP), 8 February 1986**

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Description

[0001] This invention relates to particles having surface properties and methods of making them.

[0002] It is concerned to provide surface properties on the particles that will cause them to be more firmly retained on substrates such as fabrics, vitreous surfaces or the skin. In the invention this is achieved by providing a polymer core to the particle with a surface of a different polymer.

[0003] These particles are particularly suitable for forming a stable-dispersion in a liquid concentrate containing active ingredient, such as a liquid detergent concentrate or liquid rinse conditioner containing a fragrance, wherein the active ingredient becomes imbibed or entrapped within the matrix of the polymer and this provides a product which is suitable for the controlled release of active ingredients.

Summary of the Prior Art

[0004] Various polymer systems have been proposed as diluting systems for active ingredients. WO-A-90/08478 relates to food flavouring high note enhancement by micro-bead impregnated with flavouring containing volatiles and is protected by a soluble or swellable coating. This patent describes porous polymeric beads made from copolymers of divinylbenzene with styrene. The polymeric beads are preferably produced with a coating that retards the release of the flavouring. Illustrative coatings are hydroxyl propyl methyl cellulose and polyvinyl alcohol.

[0005] EP-A-297605 is directed to cell culture microcarriers comprising (meth)acrylic ester polymer particles having positively charged groups on the surface. The polymer particles have an average diameter of between 100 and 1000 microns and are prepared by oil in water type suspension polymerisation of (meth) acrylic ester as a monomer followed by treating the resulting polymer particles with ammonia or amine having 5 or less carbon atoms. Typically polyvinyl alcohol may be used as a stabiliser in the aqueous phase of the polymer. The positive groups are generated by reacting the formed polymer particles with the amine or ammonia. In order for this to be possible it is necessary that the matrix monomer contains functional groups that react with the ammonia or amino compounds, e.g. glycidyl methacrylate.

[0006] JP-A-62289238 describes resistant anion exchangers. This patent describes an anionic exchange resin produced by polymerising glycidyl esters with other vinyl compounds and then reacting the polymer particles with ammonia or amines.

[0007] WO-A-92/18222 describes the preparation and use of solid non-porous crosslinked polymer beads. The beads are produced by first forming a monomer phase which comprises at least one monoechlenically unsaturated monomer and a monomer soluble initiator. The monomer phase is then combined with an aqueous liquid-phase which is substantially immiscible with the monomer phase to form a reaction mixture. An organic polymer colloid suspending agent can also be combined with the aqueous phase. Preferred suspending agents include polyvinyl polyols having a molecular weight of at least 40,000 and includes polyvinyl alcohol.

[0008] US-A-4842761 discloses composite particles comprising a water soluble polymer and a water insoluble polymer. Discrete entities of one polymer are embedded within a matrix of the other polymer. The composite particles are used as perfume carriers in laundry detergents and provide controlled release perfume.

[0009] GB-A-2234301 describes a method of encapsulating a water insoluble substance, such as an insecticide, within a carrier capsule. The water insoluble substance is dissolved in a substance which is capable of modifying the carrier capsule so that it imbibes said substance and then entrapping said substance within the thereby modified carrier capsule by removal of the carrier capsule modifying agent. The patent exemplifies starch granules and cellulosic fibres as carrier capsule.

[0010] EP-A-0441512 discloses a swellable oil-absorbent agent formed of a cross-linked polymer, which may absorb any of a wide range of oils, and which also provide a substrate for underwater gradual release of a chemical agent, such as a fungicide, that allows the chemical agent to dissolve out into water at a fixed speed.

[0011] EP-A-0604109 discloses a process of suspension polymerisation of water-soluble monomers in aqueous suspension to form spherical, unagglomerated polymer beads. The polymerisation is conducted in the presence of a nonionic, substituted hydroxyalkylcellulose dispersant.

[0012] WO-A-9322417 discloses a capsule for use in heavy duty liquid compositions which comprise a detergent sensitive active ingredient and a composite polymer which in turn comprises a hydrophilic polymer and a hydrophilic polymer core. The copolymer forms a network which entraps sensitive components between the hydrophobic particles and crosslinked water soluble polymers.

[0013] EP-A-285694 is one of a series of documents concerned with porous cross linked polymers obtained by polymerisation around droplets of a liquid substance which may subsequently be removed. These porous polymers can act as carriers for a variety of liquids. They are used in products for application to the skin, and give controlled release of the liquid to the skin.

[0023] It may also be possible to impart some or all of the cationic functionality to the final polymeric particles by incorporating cationic monomer units into the polyhydroxy polymer. One example of this would be to first provide a copolymer of vinyl acetate with vinyl acetamide and then hydrolyse this to a polymer which would then have vinyl alcohol units and vinyl amine units. The vinyl amine units can be rendered cationic by subsequent protonation.

[0024] The particles themselves are insoluble in water, so that they can be deposited from an aqueous wash or rinse liquor.

[0025] The functional polymer at the exterior of these particles may form, or be included within a coating or incomplete coating on these individual particles. The hydroxy functional polymer is preferably nonionic or cationic. It will be explained below that it generally constitutes between 1 and 25% of the weight of the perfume-containing particles, usually between 1 and 10%.

Particle size

[0026] Polymer particles of this invention desirably have an average particle size of at least 10 μ m, better at least 20 μ m or 30 μ m, even better at least 40 μ m, for ease of handling. The rate of release of any active ingredient in the particles may be faster than desired if the particles are of very small size such as average size of 1 μ m. The polymer particles desirably have an average size not larger than 150 μ m, better not over 125 μ m so that the particles are not easily visible after deposition.

[0027] For particles intended to be used in fabric washing or conditioning, it is especially preferred to use particles with a mean size of at least 40 μ m, e.g. 40 to 100 μ m.

[0028] For particles intended for other products, e.g. personal washing products, a faster rate of release may be desired than with fabrics products, although retention and delay in release for some hours is still desirable, consequently a smaller particle size may be advantageous, such as a mean size in the range from 10 to 50 μ m.

[0029] Polymerisation techniques generally produce a range of particle sizes. For this invention it is desirable that a high proportion of the particles lie between the above limits on particle size. Thus, when particles are intended for fabric treatment, preferably 90% or more of the particles are larger than 30 μ m. Preferably also 90% or more of the particles are not larger than 150 or even 125 μ m. Better, 95% or more are not larger than 125 or even 100 μ m.

[0030] To achieve these criteria, it may be necessary to sieve the particles and thereby separate oversized and/or undersized particles. An important aspect of reducing the amount of emulsion polymer formed (broadly speaking, polymer beads of a size below 10 μ m) is to use an initiator system that is substantially insoluble in the aqueous phase, so as to prevent initiation of monomer molecules held in micelles. Suitable water insoluble initiators include azo compounds such as azobisisobutyronitrile (AIBN) and higher alkyl peroxides.

Hydroxy functional polymer

[0031] This polymer bearing hydroxy groups and located at the exterior of the particles serves to enhance deposition onto (or retention on) skin, hair, hard surfaces especially vitreous surfaces and fabric.

[0032] This polymer is desirably such that at least 80% of the monomer residues in the polymer contain at least one free hydroxy group able to participate in hydrogen bonding. The polymer is preferably nonionic and such that, in its pure state, it is watersoluble. However, when attached to the exterior of the particles, it does not render them water soluble. It is attached in such a way that it is not washed off completely upon contact with water. In some instances there may be chemical bonding, even in a form of grafting.

[0033] The preferred polymer is selected from polymers comprising vinyl alcohol units, more preferably polyvinyl alcohol.

[0034] As is well known, polyvinyl alcohol cannot be prepared directly by polymerisation, and is obtained by the hydrolysis of polyvinyl acetate. This hydrolysis generally stops before completion, and polymers with varying amounts of hydrolysis are commercially available.

[0035] We have found that deposition of particles, especially onto fabric, is better when the polyvinyl alcohol exhibits a substantial level of hydrolysis.

[0036] We may therefore prefer to use polyvinyl alcohol which comprises of at least 80% vinyl alcohol units. More preferably we use polyvinyl alcohol which comprises of at least 85% vinyl alcohol units, i.e. in which 85 or 88 to 100% of the acetate residues have been removed by hydrolysis. Even more preferably, the polyvinyl alcohol is at least 90% better at least 96% hydrolysed, e.g. 96 to 99% hydrolysed.

[0037] Another possible polymer is a copolymer of polyvinyl alcohol, available through hydrolysis of the corresponding copolymer of vinyl acetate. For example a copolymer of vinyl acetate and vinyl formamide, with a high proportion of vinyl acetate can be hydrolysed to a copolymer of vinyl alcohol and vinyl formamide. Further hydrolysis of such a polymer converts the formamide residues to amine groups, giving a cationic copolymer of vinyl alcohol and vinyl amine.

[0038] Another category of polymers which can be used is cellulose and chemically modified cellulose where the

modification does not greatly reduce the number of hydroxyl groups present. Examples of polymers in this category are hydroxyethyl cellulose, and hydroxypropyl cellulose.

[0039] Hydroxyethyl cellulose is available commercially and is made by treatment of cellulose with sodium hydroxide, followed by reaction with ethylene oxide. Another possibility is cellulose which has been chemically modified by the introduction of cationic substituent groups.

[0040] It is a characteristic of many grades of polyvinyl alcohol and of hydroxy ethyl cellulose that free hydroxy groups constitute more than 20%, often more than 25% of the weight of the polymer.

Solid particles

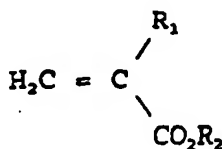
[0041] For this invention it is preferred to use polymer particles which are solid - although they may be porous as well as solid - rather than particles in the form of hollow capsules.

[0042] Advantages of solid particles are that the desired size range is accessible, and that the polymerisation reaction can be carried out in the absence of perfume.

[0043] The particles could possibly be porous particles made by polymerisation around a liquid pore-forming agent, as taught in EP-A-285694. However, we have found it to be unnecessary to include such a pore-forming agent. Polymers formed by simple polymerisation of a monomer mixture are able to absorb and carry active ingredients.

[0044] Preferred polymer particles may be formed by the polymerisation of vinyl monomers, with some cross linking and/or chain branching agent included in the monomers which are polymerised, so that some cross links are formed between the polymer chains. If a cross linking agent is used, the proportion of cross linking may be low, so that after polymerisation there may be some polymer chains which remain entirely linear and are not cross linked to any other chains.

[0045] A number of vinyl monomers containing a single carbon-carbon double bond may be used. One suitable category of monomers is the esters of acrylic and alkyl acrylic acids, of formula



where R_1 is hydrogen or alkyl of 1 to 6 carbon atoms, preferably 1 to 3 carbon atoms and R_2 is alkyl (including branched and cycloalkyl) of 1 to 7 or 8 carbon atoms, preferably 3 or 4 carbon atoms in a straight or branched chain, or 7 carbon atoms in a bridged ring.

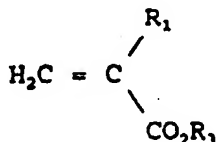
[0046] Specific examples of suitable monomers are isobutyl methacrylate (which is particularly preferred), n-butyl (meth)acrylate, isobutyl acrylate, n-propyl acrylate, iso-propylmethacrylate and norbornyl (meth)acrylate. Less preferred is methyl methacrylate.

[0047] Another suitable monomer is styrene.

[0048] It is possible to use simple linear polymers. However, these can give particles which are somewhat sticky, and it is usually convenient to introduce some cross-linking or chain branching.

[0049] Cross linking between polymer chains formed from the above monomers can be achieved by including in the monomer mixture a small proportion of a monomer with two carbon-carbon double bonds, often termed polyolefinic or multifunctional cross-linking monomers. The use of such a material to provide cross linking is well known in other applications of polymers, although it is usual to introduce a greater proportion of cross linking than is required for this invention. Examples of this type of cross linking agent are divinyl benzene, diesters formed between acrylic acid and diols and higher ester formed between acrylic acid and polyols - which may be sugars.

[0050] Chain branching can be introduced by including among the monomers a hydroxyalkyl monomer of formula



where R_1 is as specified above and R_3 is alkyl of 1 to 6 carbon atoms bearing at least one hydroxy group, preferably 3 to 4 carbon atoms in a straight or branched chain and bearing a single hydroxy group. These monomers undergo a side reaction during the course of polymerisation, and this side reaction produces chain branching. When there is chain branching without cross linking, it is suitable that hydroxyalkyl monomer of the above formula provides from 10 to 40% by weight of the monomer mixture.

[0051] Suitable hydroxyalkyl acrylate monomers are hydroxypropyl methacrylate, hydroxybutylacrylate, and hydroxyethylacrylate.

[0052] Attachment of a polymer with hydroxy groups, notably polyvinyl alcohol, at the exterior of the particles, can be achieved by polymerising the monomers in the presence of the polyvinyl alcohol (or other polymer with hydroxy groups) using the technique of suspension polymerisation.

Suspension polymerisation

[0053] Suspension polymerisation is a process in which the organic monomers are formed into a suspension in an aqueous phase, and polymerised. Polymerisation of each suspended droplet leads to a bead of polymer.

[0054] During the polymerisation process droplets of polymerising monomer and polymer particles are constantly moving through the continuous suspension liquid and statistically there is a high probability of collisions between droplets and between particles.

[0055] A method of preparing polymeric products of the present invention may be by suspension polymerisation of hydrophobic monomers or monomer blend in an aqueous medium comprising a vinyl addition cationic monomer and as a polymerisation stabiliser a polymer comprising free hydroxy groups.

[0056] A method of preparing polymer products of the present invention may be by suspension polymerisation of a blend of isobutylmethacrylate and dimethyl benzene in an aqueous medium comprising benzyl chloride quaternised dimethyl amino ethyl acrylate and as a polymerisation stabiliser a polyvinyl alcohol. It is important that agglomeration of particles is prevented as this could very quickly result in loss of stability during the polymerisation. It is customary to stabilise the suspension by incorporating a stabilising agent in the aqueous phase before adding the monomer. Polyvinyl alcohol (PVA) is known to act as a stabiliser.

[0057] Polymerisation may be carried out using a combination of PVA and second stabilising agent which may or may not be a second hydroxy polymer. It is quite feasible to use a mixture of PVA grades. Examples of materials which can be used as second stabilisation agents include other water soluble polymers such as polyacrylic acid or water soluble salts thereof. Typically the amount of stabiliser used is generally between 1 and 5% by weight of aqueous phase or 3 to 15% by weight of the monomer being polymerised, most preferably between 5 and 10% by weight of monomer.

[0058] It is possible to incorporate surfactants such as sodium oleate or sodium lauryl sulphate into the continuous aqueous phase to assist with the stabilisation of the dispersed phase monomer droplets during the polymerisation. Typically the surfactants used in this application are those which are preferentially soluble in the aqueous phase, typically surfactants which have an HLB of 8 or above. However, when choosing a surfactant in this invention it is important that it does not have an adverse effect on the droplet size and could result in the formation of unacceptably small sized polymer particles. It is particularly important to reduce the levels of emulsified droplets, which give rise to polymer particles below 10 microns. It is desirable to reduce the levels of emulsion to below 10%, particularly below 8%.

[0059] Thus, a typical polymerisation procedure will commence by forming an aqueous solution of the hydroxy functional polymer which acts as stabilising agent, together with a polymerisation initiator such as an azo or peroxide initiator, in a reaction vessel. Then while agitating the contents of the reaction vessel, the organic monomers are progressively fed in so that the monomers become dispersed in the aqueous phase and polymerise therein. The addition of monomers can be continued until the mixture in the reaction vessel is a slurry of polymer beads containing about 30% by weight of polymer.

[0060] In a possible variation of this procedure some of the monomer is dispersed in the aqueous solution of stabilising agent before any polymerisation initiator is added. In another possible variation the monomers are emulsified in water before they are added to the reaction vessel. Suspension polymerisation typically produces polymer beads with a diameter larger than 100 microns. Smaller particle sizes in the range of 50 to 100 microns can be obtained by increasing the amount of stabiliser in the aqueous phase, or by increasing the amount of agitation, or both. A method of preparing polymer particles of the instant invention, may be when the cationic vinyl addition polymer is a benzyl chloride quaternised dimethyl amino ethyl (meth)acrylate and the resulting polymer particles have an average particle size in the range 10 microns to 150 microns.

[0061] A method of preparing polymer particles of the instant invention, may be when less than 4% of the hydrophobic monomer forms polymer particles of particle size less than 10 microns. A further method of preparing polymer particles of the instant invention, may be when less than 3% of the hydrophobic monomer forms polymer particles of particle size less than 10 microns.

acid is then added to catalyse polymerisation of the dissolved urea-formaldehyde precondensate. If the solution is rapidly agitated during this polymerisation step, shells of water-insoluble, urea-formaldehyde polymer form around and encapsulate the dispersed organic mixture and molecules of the hydroxy-group containing polymer are incorporated in and at the exterior of these shells. Melanine-formaldehyde precondensate can be used in place of urea-formaldehyde, and may be preferred.

[0076] Another procedure is to form encapsulated core polymer, in the absence of perfume, and subsequently allow perfume to diffuse through the shell, into the core polymer. We have found that absorption of perfume is possible through a thin shell, even though a thicker hollow shell is capable of retaining liquid perfume. Suitably the weight of polymer forming the shell is less than the weight of polymer forming the core, and the shell to core weight ratio may lie in a range from 1:3 to 1:20, better 1:5 to 1:20.

[0077] Hydroxy functional polymer will generally provide a substantial proportion of the shell, yet constitute from 1% to 25% of the capsules.

[0078] For this procedure it has been found preferable to encapsulate monomer within an aminoplast shell, then polymerise the monomer to form a (preferably solid) core of polymer within the shell. Less preferred is to partially polymerise the core before encapsulation.

[0079] It is also possible to encapsulate a mixture of liquid monomer and fragrance, then polymerise. However, this necessarily exposes perfume to the polymerisation reaction, whereas absorption of perfume through the shell into a previously polymerised core does not.

[0080] When a product contains particles in which perfume is absorbed within polymer which is encapsulated by a thin shell, perfume can diffuse through the shell, and can be released without rupture of the shell, although the release and dispersion of perfume will be slower than for neat perfume. Thus, encapsulated polymer with absorbed perfume can provide deposition and retarded release of perfume similarly to the (preferred) arrangement when perfume is absorbed in polymer beads which have hydroxy functional polymer directly at their exterior.

[0081] Examples of preparation and use of the particles will now be given.

Example 1 (not according to the invention)

[0082] A 700ml reaction flask equipped with motorised stirrer, reflux condenser, thermometer and inlet tube for delivery from a peristaltic pump was placed in a water bath at about 65°C.

[0083] An aqueous phase was prepared by mixing hydroxyethyl cellulose (5 parts) and deionized water (168 parts). The hydroxyethyl cellulose had a degree of substitution of one and was available from Hercules Chemical Corp as NATROSOL 250L. This phase was mixed until the hydroxyethyl cellulose dissolved and was then charged into the reaction flask. Stirring was applied to the reaction flask.

[0084] A monomer phase was prepared by mixing iso-butyl methacrylate (70 parts) with a cross linking co-monomer which was 1,6-hexanedioldiacrylate (1.8 parts).

[0085] 2,2'-azo(bis)isobutyronitrile [usually abbreviated to AIBN] (2 parts) was added directly to the reaction flask and dispersed for about five minutes.

[0086] The monomer phase was added to the stirring reaction flask using a peristaltic pump over about ninety minutes. After addition the reaction mass was stirred at about 65°C for about three hours and subsequently cooled.

[0087] The polymer beads were recovered from the aqueous slurry by filtration and air dried. The beads were sieved to separate the fraction with size below 125µm.

[0088] It can be seen that in this Example, the total quantity of hydroxy-functional polymer was less than 10% by weight of monomers, and only part of that quantity becomes attached to the polymer beads.

[0089] The cross linking agent is 2% by weight and 1.63% by mole of the monomer mixture.

Example 2 (not according to the invention)

[0090] A 700ml reaction flask, equipped as in Example 1, was placed in a water bath at about 65°C.

[0091] An aqueous phase was prepared by mixing poly(vinyl alcohol) available as Gohsenol AH-22 from Nippon Gohsei British Trades and Shippers Ltd, Dagenham, Essex and having a degree of hydrolysis of 97 to 98.8% (5 parts) and deionized water (168 parts). This phase was stirred until the poly(vinyl alcohol) dissolved and was then charged into the reaction flask. Stirring was applied to the reaction flask.

[0092] A monomer phase was prepared by mixing styrene (68 parts) and 1,6-hexanedioldiacrylate (1.8 parts).

[0093] AIBN (2 parts) was added directly to the reaction flask and dispersed for about five minutes.

[0094] The monomer phase was added to the stirring reaction flask using a peristaltic pump over about ninety minutes. After addition the reaction mass was stirred at about 65°C for about three hours and subsequently cooled.

[0095] At this stage, the products was in the form of an aqueous slurry from which the polymer beads were recovered by filtration and air dried.

Examples 3 to 5 (not according to the invention)

[0096] The procedure of Example 2 was repeated with different monomers as follows:

- Example 3: Methyl methacrylate (70 parts) and 1,6-hexanedioldiacrylate (1.8 parts)
 Example 4: n-Butylmethacrylate (70 parts) and 1,6-hexanedioldiacrylate (1.8 parts)
 Example 5: iso-butylmethacrylate (54 parts) and hydroxypropylmethacrylate (18 parts)

[0097] In each of these examples the monomers were satisfactorily converted to polymer beads. These were recovered by filtration and air dried.

Example 6 (not according to the invention)

[0098] Beads, produced as in Example 5, were sieved to remove any beads larger than 75µm diameter.

[0099] A perfume was prepared consisting of equal amounts of

- i) dihydromyrcenol (2,6-dimethyl-7-octen-2-ol)
- ii) anisaldehyde
- iii) dimethylbenzylcarbonyl acetate
- (iv) 1,1,2,4,4,7-hexamethyl-6-acetyl-1,2,3,4-tetrahydronaphthalene, available under the trademark "TONALID",
- (v) 3a,4,5,6,7,7a-hexahydro-4,7-methano-1(3)H-inden-6-ylpropionate, available under the trademark "FLOROCY-CLENE",

[0100] Perfume-loaded polymer beads were prepared by mixing the above beads and perfume into a diluted rinse conditioner, to yield an aqueous slurry containing:-

Polymer beads	10.71%
Perfume	10.71%
Di-hardened tallow dimethyl ammonium chloride	3.5%
Water	Balance

[0101] This slurry was agitated for two hours and left to stand for twenty four hours, after which it appeared that all the perfume had been absorbed into the polymer beads.

[0102] This slurry was added to a quantity of a rinse conditioner formulation which was an aqueous emulsion containing a 1,2-dihardened tallowoxy-3-trimethyl ammoniopropylamine chloride (HTTMAPC) as cationic softener. This material is disclosed in US-A-4137180.

The formulation contained:

[0103]

HTTMAPC (including some fatty acid impurity)	13.5%
Ethoxylated Coconut alcohol (20EO)	0.75%
Hardened tallow alcohol	0.75%
Calcium chloride	0.2%
Preservative	0.02%
Demineralised water	Balance to 100%

[0104] After adding the slurry, the resulting perfumed formulation contained 0.75% by weight perfume, carried in polymer beads. The perfumed rinse conditioner formulation was agitated for two hours and then stored for six days in a closed container. A control formulation contained 0.75% by weight perfume, and the same concentration of fabric softener, without polymer beads.

[0105] To test perfume deposition, this rinse conditioner formulation and the control formulation were both diluted with water to provide rinse liquors containing 0.5% of the rinse conditioner formulation.

[0106] Test pieces of fabric were de-sized cotton terry towelling, approximate weight 25g. For each test, a piece of terry towelling was weighed accurately and treated with 30 times its own weight of rinse liquor, in a Tergotometer pot, stirring at 80rpm for 20 minutes. The cloth was then wrung out by hand, and line dried.

[0107] The amount of perfume in the fresh and used rinse liquors was determined by solvent extraction from 100g of rinse liquor and gas chromatographic (GC) analysis of the solvent extract. The percentage deposition of perfume materials onto the cloth was calculated for three of the five materials. The results obtained were:

Ingredient	% Deposition	
	Control	Perfume in polymer beads
Dihydromyrcenol	14	25
DMBCA	24	33
Florocyclene	38	42

[0108] The amount of perfume on the dry cloth was determined by extraction of 5g of dry cloth with 20ml ethyl acetate, followed by GC analysis of the solvent extract.

[0109] The amount of perfume detected was expressed as a percentage of the theoretical maximum quantity (which would be present if there were complete deposition onto fabric and no subsequent losses).

[0110] The results obtained for materials were:

Ingredient	% Ingredient remaining on dry cloth	
	Control	Perfume in polymer beads
Dihydromyrcenol	not detected	1.3
DMBCA	not detected	8.0
Florocyclene	not detected	9.8
Tonalid 2	30.2*	51.1*

* = result possibly affected by other GC peaks.

Example 7 (not according to the invention)

[0111] Perfume mixture was incorporated into polymer beads, as set out in the previous Example, but with two ratios of polymer to perfume. These were 1:1 polymer to perfume and 1:2 polymer to perfume.

[0112] The resulting perfumed beads were incorporated into rinse conditioner formulations as in the previous example so as to provide 0.75% by weight perfume in each formulation. A control formulation contained 0.75% by weight perfume, but no polymer. These were diluted to rinse liquors containing 0.5% by weight fabric conditioner, and used to treat terry towelling as in the previous Example.

[0113] The treated cloths were assessed by a panel of eight people.

[0114] Assessments were made on damp cloth directly after treatment, on dry cloth 24 hours after treatment and on dry cloth five days after treatment.

[0115] The assessments were:

Assessment Stage	Control (no polymer)	1:1 polymer to perfume	1:2 polymer to perfume
Damp Cloth	Strongest	Weaker than control Equal to (1:2)	Weaker than control Equal to (1:1)
Dry Cloth (24 hours)	Intense Tonalid odour	Florocyclene, Tonalid and anisaldehyde odours Stronger than (1:2)	Florocyclene, Tonalid and anisaldehyde odours Weaker than (1:1)
Dry Cloth (five days)	Tonalid odour	Florocyclene, Tonalid and anisaldehyde odours Weaker than (1:2)	Florocyclene, Tonalid and anisaldehyde odours Stronger than (1:1)

Example 8 (not according to the invention)

[0116] Beads were produced as in Example 2. The monomer mixture was isobutyl methacrylate (70 parts) mixed with 1,6-hexanedioldiacrylate (1.8 parts). Preparations were carried out using various different grades of polyvinyl alcohol and some other materials as the stabiliser. The grades of polyvinyl alcohol differed in the extent to which they had been hydrolysed from polyvinyl acetate.

[0117] The materials used were:

	Stabiliser	Hydrolysis	Tradename
A	polyvinyl alcohol	98-99%	Gohsenol N-300
B	polyvinyl alcohol	97-98.5%	Gohsenol A-300
C	polyvinyl alcohol	87-89%	Gohsenol GH-23
D	polyvinyl alcohol	78-80%	Gohsenol KH-17
E	polyvinyl alcohol jointly with a surface active acrylamide copolymer	97-99%	Gohsenol AH-22
F	polyvinyl alcohol jointly with a 95:5 copolymer of acrylamide and behenyl (25 ethylene oxide) methacrylate	97-99%	Gohsenol AH-22
G	hydroxyethyl-cellulose		Natrosol 250L
H	surface active acrylamide copolymer (comparative example)		MER 10

[0118] In all cases the dried beads were mixed with their own weight of a single perfumery material, Florocyclene, also used in Example 6. After mixing they were left to equilibrate overnight, then a small quantity of finely divided silica (2% based on the total weight of perfume and polymer) was added as a surface improving flow aid.

[0119] The perfume-containing polymer beads were mixed into rinse conditioner (as used in Example 6) using a magnetic stirrer. In each case the amount of perfume carrying beads was 1.5% by weight of the rinse conditioner, so that the quantity of perfume was 0.75% by weight of the rinse conditioner formulation. A control formulation was prepared containing 0.75% Florocyclene without the polymer.

[0120] Deposition onto fabric was assessed using test cloths made of cotton terry towelling, acrylic fabric, and polyester. The fabric washing procedure was the same as set out in Example 6. After treatment of the fabric and drying, the intensity of Florocyclene on groups of the dried test cloths was assessed by a panel of five assessors who ranked the cloths in order of intensity of Florocyclene. The following results were obtained as unanimous views of the assessors:

On cotton test pieces B>A>C>Control
 C>D>Control
 E>F>Control
 and G>H>Control
 On acrylic test pieces A>B>C>Control

[0121] On polyester test pieces A,B and C were all stronger than the control, but the assessors were not unanimous as to which of the three gave the highest perfume intensity.

[0122] These results demonstrate that the polymer particles A,B,C and D made using polyvinyl alcohol as stabiliser and hydroxy-functional monomer all gave an increase in the perceived intensity of perfume on fabric. So did polymer particles E and F made using polyvinyl alcohol in a mixture of stabilisers. The polymer particles G made using hydroxyethylcellulose as stabiliser and hydroxy-functional monomer also gave an increase in the perceived intensity of perfume on fabric and were superior to the particles H made using a different polymer as stabiliser.

Example 9

[0123] 942 parts of water, 1228 parts of a 10% solution of 88% converted polyvinyl alcohol, 4 parts benzyl chloride quaternised dimethylaminoethyl methacrylate (MADQUAT) and 6 parts Tegofomex defoamer were charged to a round bottomed reactor vessel. The stirrer was started at 150rpm and a nitrogen bleed was applied and the contents of the vessel were warmed to between 60 and 62°C. 12.3 parts of Vaso 67 thermal initiator slurried in 36 parts of water was then added to the contents of the vessel. A monomer blend comprising 1225 parts of isobutyl methacrylate and 24.5 parts butane diol diacrylate were then charged to the reactor over 1 hour to provide reaction solids of 35%. The tem-

perature of the reaction needed was held at 60 - 62°C until monomer addition was completed and then the temperature of the was allowed to rise under the reaction exotherm to 80°C where it was held for 3 hours. The reaction medium was then cooled and the particle size range of the polymer was established as 40 - 900 microns with a means of 81.1 microns. The composition contained less than 9% emulsion, i.e. of beads less than 10µm.

Example 10

[0124] Example 9 was repeated but in a flat bottom vessel with a stirrer speed of 300rpm and using lauryl peroxide as the initiator. The particle size range was from 20-200 microns, with a mean particle size of 79 microns. The composition contained less than 8% emulsion.

Example 11

[0125] Example 9 was repeated by varying the amount of hydrophobic monomer blend charged to the reactor so as to provide reaction solids of 30%, 35% and 40%. The results are as follows:-

Solids	Particle Size Range of Beads	Emulsion
30%	40-350 microns	27.7%
40%	10-200 microns	24.6%

Example 12

[0126] Example 9 was repeated at 25% solids content but varying the amounts of stabiliser, defoamer and cationic monomer.

% Stabiliser	Additions	Particle Size Range	Emulsion
10% PVA	2% Lauryl bromide quaternised dimethyl aminoethyl methacrylate dimethyl	V.thick bimodal 6-200 microns	not measured
5% PVA	2% Arquad T-30	Small lumps aggregated beads	not measured
5% PVA	2% Disponil A3065	V.Broad distribution	not measured
10% PVA	1% NaCl and 5% Methyl end capped PEG 2000	40-400 microns	46.3%
10% PVA	1% NaCl 5% MPEG 1000	Bimodal 40-400 microns	not measured
10% PVA	1% NaCl 10% MPEG 2000	40-150 microns	36.8%
10% PVA	1% NaCl 5% Methyl end capped PEG 2000 2% Synperonic PE/L61	40-400 microns	43.2%

Example 13

[0127] Example 9 was repeated using an amount of hydrophobic monomer that would provide a reaction medium of 25% solids and different grades of polyvinyl alcohol were used having different degrees of hydrolysis. In one test sodium chloride was added to the aqueous phase and the tests using 98% hydrolysed PVA, lauryl bromide quaternised dimethylaminoethyl methacrylate was used in place of the benzyl chloride quaternary cationic monomer.

% Stabiliser	Stabiliser type	Additions	Results
10%	88% hydrolysed (high mol wt)		100-200 microns
10%	73% hydrolysed		100-200 microns
10%	88% hydrolysed (high mol wt)	1% NaCl	Large beads formed
5%	98% hydrolysed	0.15% Lauryl Bromide Quat	Polymer turned solid
5% A-300	98% hydrolysed	0.2% Lauryl Bromide Quat	Large beads formed
10% A-300	98% hydrolysed	0.5% Lauryl Bromide Quat	Very small beads formed

Example 14

[0128] Example 9 was repeated but using different levels of stabiliser.

% Stabiliser	Results
15%	0.2 - 150 microns, mean 13 microns
10%	20 - 400 microns, mean 109 microns
7.5%	20 - 400 microns, mean 122 microns
5%	20 - 900 microns, mean 105 microns
2.5%	60 - 800 microns, mean 222 microns

Example 15

[0129] Example 9 was repeated using the following cationic monomers:-

Product	Cationic Monomer Used
A	tertiary butyl aminoethylmethacrylate (TBAEMA)
B	Methyl chloride quaternised DMAEMA (MBJ)
C	Dimethyl aminopropyl methacrylamide (DMPMA)
D	Benzyl chloride quaternised DMAEMA (MADQUAT BZ 75)

In performance tests the products A-D were dispersed into a perfume base having a cationic character to assess whether the particles form agglomerates. Products A and B formed agglomerates, product C formed some agglomerates but D was almost free of agglomerates.

Example 16

[0130] Example 9 was repeated using DMAPMA acetate salts, with different degrees of neutralisation, as the cationic monomer component. The amount of emulsion produced was measured.

Additions	Results	Emulsion
2% DMAPMA 100% neutralised	pH 5.42	33%
2% DMAPMA 75% neutralised	pH 8.0	28.7%
2% DMAPMA 50% neutralised	pH 8.4	18.9%
2% DMAPMA 25% neutralised	pH 8.6	19.6%

Example 17

[0131] Example 9 was repeated but using various levels of benzyl chloride quaternised dimethylaminoethyl methacrylate.

Amount of cationic monomer	Performance test in perfume base	Emulsion
1.0%	no agglomerates occurred	16.2%
0.5%	no agglomerates occurred	13.0%
0.2%	virtually none	17.9%
0.1%	formed small white agglomerates	15.8%

Example 18

[0132] The polymer particles produced in Example 9 evaluated by assessing the ability to absorb water insoluble active ingredients. 10 parts of beads were dispersed into 10 parts of liquid water insoluble active ingredients and the quantity of liquid absorbed was observed visually.

Active	Description	Volume pre Absorption	% Absorption
Propischlor	Herbicide	1	0
Gossyplure	Semio Chemical	1.2	100
Methyl Eugenol	Semio Chemical	1	100
Trimedlure	Semio Chemical	1	50
Spiroketal	Semio Chemical	1	100
Pyrethrum Extract 25% Active in paraffinic solvent	Insecticide	1	20
Pyrethrum Extract 50% Active	Insecticide	1.1	50
Ethion	Insecticide	1	50
Cypermethrin 50% Active in Solvesso 200 solvent	Insecticide	2	50
Furfural	Soil Sterilant	1	100

Example 19

[0133] A preparation of isobutyl methacrylate polymer was repeated with differing quantities of MADQUAT as the only variable. The results show that mean particle size increases with quantity of such a cationic monomer.

% MADQUAT	Mean particle size, μm
0.22	74
0.3	98
0.4	109
0.5	165

Claims

1. Water-insoluble organic polymer particles formed of hydrophobic core polymer and having at their surface at least one further polymer which contributes a hydroxyl and a cationic functionality at that surface, wherein the cationic functionality is provided by pendant cationic groups that have been derived from cationic vinyl addition monomer units bonded to the organic water-insoluble polymer or cationic monomer units present in the further free hydroxy containing polymer.
2. Polymeric particles according to claim 1 wherein the particles have an average particle size of at least 40 micrometer (microns) but no more than 150 micrometer (microns).
3. Polymeric particles according to any preceding claim wherein cationic functionality at the surface is provided by one or more cationic vinyl addition monomers selected from quaternised salts of dimethyl amino propyl acrylamide and dimethyl amino propyl methacrylamide.
4. Polymeric particles according to any preceding claim wherein cationic functionality at the surface is provided by one or more cationic vinyl addition monomers selected from $\text{C}_4\text{-C}_8$ alkyl, aryl, alkaryl, aralkyl halide of dimethyl amino (meth)acrylates.
5. Polymeric particles according to any preceding claims wherein cationic functionality at the surface is provided by a benzyl-chloride-quaternised dimethyl amino ethyl methacrylate monomer.

6. Polymeric particles according to any preceding claim wherein the core polymer is based on alkyl esters of acrylic acid or methacrylic acid.
7. Polymeric particles according to any preceding claims wherein the core polymer is based on alkyl ester of acrylic acid or methacrylic acid and a multifunctional cross-linking monomer.
8. Polymeric particles according to any preceding claim wherein the core polymer is based on isobutyl methacrylate and divinyl benzene.
9. Polymeric particles according to any preceding claim wherein the further polymer providing hydroxy functionality is selected from polymers comprising vinyl alcohol units.
10. Polymeric particles according to any preceding claim wherein the polymer providing hydroxy functionality is selected from polyvinyl alcohol comprising at least 80% vinyl alcohol units.
11. Polymeric particles according to any preceding claim wherein the polymer providing hydroxy functionality is selected from polyvinyl alcohol comprising at least 85% vinyl alcohol units.
12. Polymeric particles according to any preceding claims comprising an active ingredient to be released in a controlled manner, wherein the active ingredient is selected from insecticides, curing agents and sunscreen agents.
13. Polymeric particles according to any preceding claim which form stable dispersions in liquid medium containing active ingredients and remain substantially free from agglomerated particles.
14. Polymeric particles according to claim 13 which are capable of imbibing and entrapping the active ingredient, whilst remaining in suspension and substantially free from agglomerated particles.
15. A method of preparing polymeric products according to any preceding claim by suspension polymerisation of hydrophobic monomers or monomer blend in an aqueous medium comprising a vinyl addition cationic monomer and as a polymerisation stabiliser a polymer comprising free hydroxy groups.
16. A method of preparing polymer products according to any preceding claims by suspension polymerisation of a blend of isobutylmethacrylate and dimethyl benzene in an aqueous medium comprising benzyl chloride quaternised dimethyl amino ethyl acrylate and as a polymerisation stabiliser a polyvinyl alcohol.
17. A method of preparing polymer particles according to claim 16 or claim 17 wherein the suspension polymerisation process is effected by azo or peroxide initiators.
18. A method of preparing polymer particles according to any preceding claim wherein the cationic vinyl addition monomer is a benzyl chloride quaternised dimethyl amino ethyl (meth)acrylate and the resulting polymer particles have an average particle size in the range 10 micrometer (microns) to 150 micrometer (microns).
19. A method of preparing polymer particles according to any preceding claim wherein less than 4% of the hydrophobic monomer forms polymer particles of particle size less than 10 micrometer (microns).
20. A method of preparing polymer particles according to any preceding claim wherein less than 3% of the hydrophobic monomer forms polymer particles of particle size less than 10 micrometer (microns).

Patentansprüche

1. Wasserunlösliche organische Polymerteilchen, gebildet aus einem hydrophoben Kernpolymeren, die an ihrer Oberfläche mindestens ein weiteres Polymeres aufweisen, das an der Oberfläche Hydroxylgruppen und kationische funktionelle Gruppen bereitstellt, wobei die kationischen funktionellen Gruppen durch seitenständige kationische Gruppen bereitgestellt werden, die sich von kationischen Vinyladditions-Monomereinheiten, die an das organische wasserunlösliche Polymere gebunden sind, oder von kationischen Monomereinheiten, die im weiteren, freie Hydroxylgruppen enthaltenden Polymeren vorhanden sind, ableiten.

2. Polymerteilchen nach Anspruch 1, wobei die Teilchen eine durchschnittliche Teilchengröße von mindestens 40 µm, jedoch nicht mehr als 150 µm aufweisen.
- 5 3. Polymerteilchen nach einem der vorstehenden Ansprüche, wobei die kationischen funktionellen Gruppen an der Oberfläche durch ein oder mehr kationische Vinyladditionsmonomere bereitgestellt werden, die unter quaternisierten Salzen von Dimethylaminopropylacrylamid und Dimethylaminopropylmethacrylamid ausgewählt sind.
- 10 4. Polymerteilchen nach einem der vorstehenden Ansprüche, wobei die kationischen funktionellen Gruppen an der Oberfläche durch ein oder mehr kationische Vinyladditionsmonomere bereitgestellt werden, die unter C₄-C₈-Alkyl-, Aryl-, Alkaryl- oder Aralkylhalogeniddimethylamino(meth)acrylaten ausgewählt sind.
- 15 5. Polymerteilchen nach einem der vorstehenden Ansprüche, wobei die kationischen funktionellen Gruppen an der Oberfläche durch ein mit Benzylchlorid quaternisiertes Dimethylaminoethylmethacrylat-Monomeres bereitgestellt werden.
- 20 6. Polymerteilchen nach einem der vorstehenden Ansprüche, wobei das Kernpolymere auf Alkylestem von Acrylsäure oder Methacrylsäure basiert.
7. Polymerteilchen nach einem der vorstehenden Ansprüche, wobei das Kernpolymere auf Alkylestem von Acrylsäure oder Methacrylsäure und einem multifunktionellen vernetzenden Monomeren basiert.
- 25 8. Polymerteilchen nach einem der vorstehenden Ansprüche, wobei das Kernpolymere auf Isobutylmethacrylat und Divinylbenzol basiert.
- 29 9. Polymerteilchen nach einem der vorstehenden Ansprüche, wobei das weitere Polymere, das funktionelle Hydroxylgruppen bereitstellt, unter Polymeren, die Vinylalkoholeinheiten umfassen, ausgewählt ist.
- 30 10. Polymerteilchen nach einem der vorstehenden Ansprüche, wobei das Polymere, das funktionelle Hydroxylgruppen bereitstellt, unter Polyvinylalkoholen, die mindestens 80 % Vinylalkoholeinheiten umfassen, ausgewählt ist.
- 35 11. Polymerteilchen nach einem der vorstehenden Ansprüche, wobei das Polymere, das funktionelle Hydroxylgruppen bereitstellt, unter Polyvinylalkoholen, die mindestens 85 % Vinylalkoholeinheiten umfassen, ausgewählt ist.
12. Polymerteilchen nach einem der vorstehenden Ansprüche, umfassend einen in kontrollierter Weise freizusetzenden Wirkstoff, wobei der Wirkstoff unter Insektiziden, Härtungsmitteln und Sonnenschutzmitteln ausgewählt ist.
- 35 13. Polymerteilchen nach einem der vorstehenden Ansprüche, die in einem flüssigen Medium, das Wirkstoffe enthält, stabile Dispersionen bilden und im wesentlichen frei von agglomerierten Teilchen bleiben.
- 40 14. Polymerteilchen nach Anspruch 13, das zum Aufsaugen und Einfangen der Wirkstoffe befähigt ist, während es in Suspension und im wesentlichen frei von agglomerierten Teilchen bleibt.
- 45 15. Verfahren zur Herstellung von Polymerprodukten nach einem der vorstehenden Ansprüche durch Suspensionspolymerisation von hydrophoben Monomeren oder Monomergemischen in einem wässrigen Medium, das kationische Vinyladditionsmonomere und als Polymerisationsstabilisator ein Polymeres, das freie Hydroxylgruppen enthält, umfasst.
- 50 16. Verfahren zur Herstellung von Polymerprodukten nach einem der vorstehenden Ansprüche durch Suspensionspolymerisation eines Gemisches aus Isobutylmethacrylat und Dimethylbenzol in einem wässrigen Medium, das mit Benzylchlorid quaternisiertes Dimethylaminoethylacrylat und als Polymerisationsstabilisator einen Polyvinylalkohol umfasst.
- 55 17. Verfahren zur Herstellung von Polymerteilchen nach Anspruch 16 oder 17, wobei das Verfahren der Suspensionspolymerisation durch Azo- oder Peroxid-Initiatoren herbeigeführt wird.
18. Verfahren zur Herstellung von Polymerteilchen nach einem der vorstehenden Ansprüche, wobei es sich beim kationischen Vinyladditionsmonomeren um ein mit Benzylchlorid quaternisiertes Dimethylaminoethyl(meth)acrylat handelt und die erhaltenen Polymerteilchen eine durchschnittliche Teilchengröße im Bereich von 10 µm bis 150

! tout en demeurant en suspension, et qui sont essentiellement exemptes de particules agglomérées.

- 5 15. Procédé de préparation de produits polymères selon l'une quelconque des revendications précédentes par polymérisation en suspension de monomères hydrophobes ou de mélange de monomères dans un milieu aqueux comprenant un monomère d'addition vinylique cationique et en tant que stabilisant de polymérisation, un polymère comprenant des groupes hydroxy libres.
- 10 16. Procédé de préparation de produits polymères selon l'une quelconque des revendications précédentes par polymérisation en suspension d'un mélange de méthacrylate d'isobutyle et de diméthylbenzène dans un milieu aqueux comprenant le chlorure de benzyle quaternisé acrylate de diméthylaminoéthyle et en tant que stabilisant de polymérisation, un alcool polyvinylique.
- 15 17. Procédé de préparation de particules polymères selon la revendication 16 ou la revendication 17, où le procédé de polymérisation en suspension est amorcé par des amorceurs azoïques ou peroxydiques.
- 20 18. Procédé de préparation de particules polymères selon l'une quelconque des revendications précédentes, où le monomère cationique d'addition vinylique est un chlorure de benzyle quaternisé (méth)acrylate de diméthylaminoéthyle et les particules polymères résultantes présentent une taille médiane des particules qui se trouve dans le domaine de 10 micromètres (microns) à 150 micromètres (microns).
- 25 19. Procédé de préparation de particules polymères selon l'une quelconque des revendications précédentes, où moins de 4 % du polymère hydrophobe forment des particules polymères d'une taille inférieure à 10 micromètres (microns).
- 30 20. Procédé de préparation de particules polymères selon l'une quelconque des revendications précédentes, où moins de 3 % du monomère hydrophobe forment des particules polymères d'une taille des particules inférieure à 10 micromètres (microns).